# SYNTHESIS AND ANTIBACTERIAL PROPERTIES OF 7-[2-(3-SUBSTITUTED-5-ISOXAZOLYL)-2-METHOXYIMINOACETAMIDO]CEPHALOSPORANIC ACID DERIVATIVES

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The synthesis of new 2-(3-substituted-5-isoxazolyl)-2-methoxyiminoacetic acids and their condensation derivatives with a suitable cephalosporanic nucleus, is reported. Their antibacterial properties were tested *in vivo* and *in vitro* also against  $\beta$ -lactamase producer microorganisms; particularly the oral bioavailability of some of these new derivatives was studied.

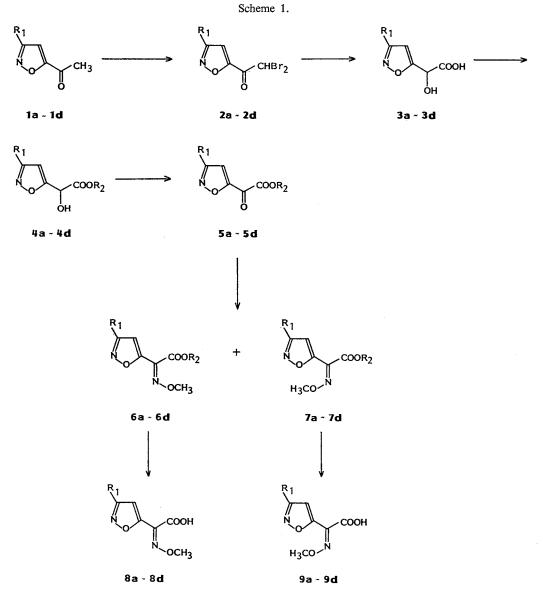
In the past few years many molecules in which a suitable cephalosporanic nucleus is condensed with alkoxyiminoacetic acid derivatives bearing a heterocycle such as furane or 2-aminothiazole<sup>1)</sup>, have been extensively studied in order to obtain new antibacterial drugs, especially those active against  $\beta$ -lactamase producer bacteria. However, these broad-spectrum antibiotics have the disadvantage of requiring parenteral administration; therefore the possibility of increasing their oral absorbtion remains one of the most important goals in chemotherapy.

We were interested to determine if the choice of an isoxazole ring as the heteroaromatic moiety could improve the oral bioavailability as previous proved for the penicillin series<sup>2</sup>). As part of our investigations in the isoxazole field<sup>3,4</sup>, we developed the synthesis of new 2-(3-substituted-5-isoxazolyl)-2-methoxyiminoacetic acids  $8a \sim 8d$  and  $9a \sim 9d$  for their use in synthesizing new cephalosporin derivatives potentially active in chemotherapy. Thus we prepared derivatives 10a, 10b, 11a ~ 11d, 12a and 12c and investigated the structure-activity relationships regarding their antibacterial properties; we were also interested in evaluating a possible improvement of their oral bioavailability compared with cefuroxime.

#### Chemistry

The synthesis of 2-(3-substituted-5-isoxazolyl)-2-methoxyiminoacetic acids  $8a \sim 8d$  and  $9a \sim 9d$ , key intermediates for the preparation of cephalosporanic derivatives, is reported in Scheme 1. The synthesis was carried out using known carbonyl compounds  $1a \sim 1d^{3,5,6}$  as starting materials.

The reaction between derivatives  $1a \sim 1d$  and bromine in acetic acid gave dibrominated compounds  $2a \sim 2d$  which were treated without any purification with aqueous sodium hydroxide in ethanol affording  $\alpha$ -hydroxyacids  $3a \sim 3d$ . These unstable compounds were directly transformed into corresponding esters  $4a \sim 4d$  which were oxidized with chromic anhydride in acetic acid to give  $\alpha$ -keto esters  $5a \sim 5d$ . Analytical data of derivatives  $4a \sim 4d$  and  $5a \sim 5d$  are reported in Table 1. The reaction of intermediates  $5a \sim 5d$  and methoxyamine hydrochloride in aqueous alcohol yielded the mixture of (Z)- and (E)-oximes which were easily separated by column chromatography. The study of the



<sup>1</sup>H NMR spectra made possible the assignment of the (Z)-isomerism to compounds  $6a \sim 6d$  and the (E)-isomerism to compounds  $7a \sim 7d$  in agreement with the previously reported literature data<sup>7</sup>). Analytical data of derivatives  $6a \sim 6d$  and  $7a \sim 7d$  are described in Table 2. A mild basic hydrolysis in aqueous ethanol of compounds  $6a \sim 6d$  and  $7a \sim 7d$  gave respectively (Z)-methoxyiminoacetic acids  $8a \sim 8d$  and (E)-methoxyiminoacetic acids  $9a \sim 9d$ . Analytical data of key intermediates  $8a \sim 8d$  and  $9a \sim 9d$  are reported in Table 3.

The synthesis of 7-aminodesacetoxycephalosporanic acids 10a, 10b and 7-aminocephalosporanic acid  $11a \sim 11d$  is described in Scheme 2. Acid chlorides of compounds  $8a \sim 8d$ , generated *in situ* with oxalyl chloride in mild conditions, reacted with silylated cephalosporanic intermediates to afford desired 7-aminodesacetoxycephalosporanic acids 10a, 10b and 7-aminocephalosporanic acids  $11a \sim 11d$ . Analytical data of compounds 10a, 10b and 11a  $\sim 11d$  are reported in Table 4. The synthesis

Table 1. 2-(3-Substituted-5-isoxazolyl)-2-hydroxyacetic esters  $4a \sim 4d$  and 2-(3-substituted-5-isoxazolyl)-2-oxoacetic esters  $5a \sim 5d$ .

Com- pound	<b>R</b> <sub>1</sub>	R₂	Yield (%)	MP <sup>a</sup> (°C) or bp (°C/Torr)	Molecular formula <sup>b</sup>	<sup>1</sup> H NMR (CDCl <sub>3</sub> ) $\delta$				
<b>4</b> a	Br	$C_2H_5$	65	50~51	C <sub>7</sub> H <sub>8</sub> BrNO <sub>4</sub>	1.31 (3H, t), 4.37 (2H, q), 5.40 (1H, d), 6.50 (1H, s)				
4b	$C_6H_5$	$C_2H_5$	63	82~83	$C_{13}H_{13}NO_4$	1.22 (3H, t), 4.26 (2H, q), 5.56 (1H, d), 7.15 (1H, s), 7.40~8.20 (5H, m)				
4c	$CH_3$	$\mathrm{CH}_3$	55	130/0.15	$C_7H_9NO_4$	2.24 (3H, s), 3.78 (3H, s), 5.42 (1H, d), 6.40 (1H, s)				
4d	OCH <sub>3</sub>	$\mathrm{CH}_3$	66	130/0.2	$C_7H_9NO_5$	3.88 (3H, s), 4.00 (3H, s), 5.30 (1H, d), 6.04 (1H, s)				
5a	Br	$C_2H_5$	80	115/0.3	C7H6BrNO4	1.47 (3H, t), 4.50 (2H, q), 7.54 (1H, s)				
5b	$C_6H_5$	$C_2H_5$	91	41~42	$C_{13}H_{11}NO_4$	1.46 (3H, t), 4.50 (2H, q), 7.50~8.20 (6H, m)				
5c	$CH_3$	$CH_3$	70	82~83	$C_7H_7NO_4$	2.46 (3H, s), 4.04 (3H, s), 7.37 (1H, s)				
5d	$OCH_3$	$CH_3$	71	<b>97~9</b> 8	$C_7H_7NO_5$	4.01 (3H, s), 4.08 (3H, s), 7.03 (1H, s)				

<sup>a</sup> Crystallization solvent: Isopropyl ether.

<sup>b</sup> Microanalyses were in satisfactory agreement with calculated values: C  $\pm$ 0.32, H  $\pm$ 0.18, N  $\pm$ 0.20.

Table 2. 2-(3-Substituted-5-isoxazolyl)-(Z)-2-methoxyiminoacetic esters  $6a \sim 6d$  and 2-(3-substituted-5-isoxazolyl)-(E)-2-methoxyiminoacetic esters  $7a \sim 7d$ .

Com- pound	$\mathbf{R}_1$	$R_2$	Yield (%)	MP <sup>a</sup> (°C) or bp (°C/Torr)	Molecular formula <sup>b</sup>	<sup>1</sup> H NMR (CDCl <sub>3</sub> ) $\delta$
6a	Br	$C_2H_5$	35	135/0.2	$C_8H_9BrN_2O_4$	1.41 (3H, t), 4.15 (3H, s), 4.21 (2H, q), 6.67 (1H, s)
6b	$C_6H_5$	$C_2H_5$	36	44~45	$C_{14}H_{14}N_{2}O_{4}$	1.40 (3H, t), 4.13 (3H, s), 4.47 (2H, q), 6.93 (1H, s), 7.30~8.10 (5H, m)
6c	$\mathrm{CH}_3$	$CH_3$	39	22~23	$C_8H_{10}N_2O_4$	2.37 (3H, s), 3.98 (3H, s), 4.15 (3H, s), 6.47 (1H, s)
6d	$\rm OCH_3$	$\mathrm{CH}_{3}$	33	165/0.1	$C_8H_{10}N_2O_5$	3.98 (3H, s), 4.05 (3H, s), 4.13 (3H, s), 6.24 (1H, s)
7a	Br	$C_2H_5$	56	140/0.15	$\mathbf{C_8H_9BrN_2O_4}$	1.41 (3H, t), 4.24 (3H, s), 4.43 (2H, q), 7.07 (1H, s)
7b	$C_{\theta}H_{5}$	$C_2H_5$	57	46~47	$C_{14} {\rm H}_{14} {\rm N}_2 {\rm O}_4$	1.40 (3H, t), 4.28 (3H, s), 4.50 (2H, q), 7.42 (1H, s), 7.40~8.10 (5H, m)
7c	$\mathrm{CH}_3$	$\mathrm{CH}_3$	50	59 ~ 60	$C_8H_{10}N_2O_4$	2.38 (3H, s), 3.98 (3H, s), 4.24 (3H, s), 6.91 (1H, s)
7d	$OCH_3$	$\mathrm{CH}_3$	58	69 <b>~</b> 70	$\mathbf{C}_8\mathbf{H}_{10}\mathbf{N}_2\mathbf{O}_5$	3.96 (3H, s), 4.05 (3H, s), 4.23 (3H, s), 6.65 (1H, s)

<sup>a</sup> Crystallization solvent: Isopropyl ether.

<sup>b</sup> Microanalyses were in satisfactory agreement with calculated values: C  $\pm 0.28$ , H  $\pm 0.15$ , N  $\pm 0.18$ .

of 7-aminocephalosporanic acids 12a, 12c is described in Scheme 3. The reaction was carried out in the same mild conditions indicated for (Z)-isomer. Analytical data of compounds 12a, 12c are reported in Table 4.

### Biological Results and Discussion

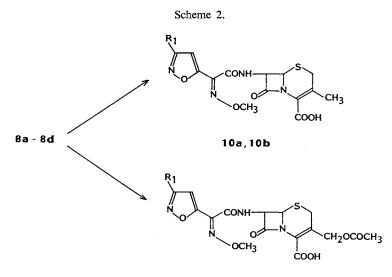
The *in vitro* antibacterial activities of 7-aminodesacetoxycephalosporanic acid derivatives 10a, 10b and 7-aminocephalosporanic acid derivatives  $11a \sim 11d$ , 12a and 12c are shown in Table 5. Cefuroxime was used as a reference compound. According to the data obtained by the *in vitro* studies, the substitution of the heterocyclic moiety such as furane, with an isoxazole ring gave derivatives

Com- pound	K.		MP (°C) (crystallization solvent) <sup>a</sup>	Molecular formula <sup>b</sup>	<sup>1</sup> H NMR (solvent) $\delta$				
	Br	94	136~137 (A)	C <sub>6</sub> H <sub>5</sub> BrN <sub>2</sub> O <sub>4</sub>	(DMSO-d <sub>6</sub> ), 4.05 (3H, s), 7.32 (1H, s)				
8b	$C_6H_5$	92	156~157 (B)	$C_{12}H_{10}N_2O_4$	(DMSO- $d_6$ ), 4.10 (3H, s), 7.50~6.30 (6H, m)				
8c	$CH_3$	94	147~148 (C)	$C_7H_8N_2O_4$	$(DMSO-d_6)$ , 2.31 (3H, s), 4.05 (3H, s), 6.79 (1H, s)				
8d	$\mathrm{OCH}_3$	96	147~148 (A)	$\mathbf{C_7H_8N_2O_5}$	(CD <sub>3</sub> OD), $4.02$ (3H, s), $4.12$ (3H, s), 6.27 (1H, s)				
9a	Br	95	141~142 (A)	$C_6H_5BrN_2O_4$	$(DMSO-d_6)$ , 4.20 (3H, s), 7.48 (1H, s)				
9b	$\mathrm{C}_{6}\mathrm{H}_{5}$	94	108~109 (A)	$C_{12}H_{10}N_2O_4$	(DMSO- $d_{\theta}$ ), 4.23 (3H, s), 7.50~8.30 (6H, m)				
9c	CH <sub>3</sub>	97	134~135 (C)	$\mathbf{C_7H_8N_2O_4}$	$(DMSO-d_8)$ , 2.34 (3H, s), 4.14 (3H, s), 7.03 (1H, s)				
9d	OCH <sub>3</sub>	98	121~122 (A)	$C_7H_8N_2O_5$	(CD <sub>3</sub> OD), 4.00 (3H, s), 4.21 (3H, s), 6.62 (1H, s)				

Table 3. 2-(3-Substituted-5-isoxazolyl)-(Z)-2-methoxyiminoacetic acids  $8a \sim 8d$  and 2-(3-substituted-5-isoxazolyl)-(E)-2-methoxyiminoacetic acids  $9a \sim 9d$ .

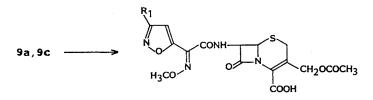
<sup>a</sup> Crystallization solvent: (A) Benzene, (B) 1,2-dichloroethane, (C) isopropyl ether.

<sup>b</sup> Microanalyses were in satisfactory agreement with calculated values: C  $\pm$ 0.31, H  $\pm$ 0.18, N  $\pm$ 0.17.



11a ~ 11**d** 





12a,12c

Com- pound	R <sub>1</sub>	Yield (%)	Dec point <sup>a</sup> (°C)	Molecular formula <sup>b</sup>	<sup>1</sup> H NMR (DMSO- $d_{\theta}$ ) $\delta$ , typical signals
10a	Br	81	71~73	$C_{14}H_{13}BrN_4O_6S$	2.00 (3H, s, CH <sub>3</sub> ), 4.05 (3H, s, CH <sub>3</sub> ON), 7.15 (1H, s, isoxazol), 10.00 (1H, d, CONH)
10b	$C_6H_5$	58	92~94	$C_{20}H_{18}N_4O_6S$	2.07 (3H, s, CH <sub>3</sub> ), 4.07 (3H, s, CH <sub>3</sub> ON), 7.40 (1H, s, isoxazol), 10.10 (1H, d, CONH)
<b>11a</b>	Br	68	70~72	$C_{16}H_{15}BrN_4O_8S$	2.04 (3H, s, CH <sub>3</sub> CO), 4.04 (3H, s, CH <sub>3</sub> ON), 7.13 (1H, s, isoxazol), 10.00 (1H, d, CONH)
11b	$C_6H_5$	46	75 <b>~</b> 77	$C_{22}H_{20}N_4O_8S$	2.08 (3H, s, CH <sub>3</sub> CO), 4.07 (3H, s, CH <sub>3</sub> ON), 7.42 (1H, s, isoxazol), 10.10 (1H, d, CONH)
11c	CH <sub>3</sub>	73	80~82	$C_{17}H_{18}N_4O_8S$	2.05 (3H, s, CH <sub>3</sub> CO), 2.28 (3H, s, CH <sub>3</sub> ), 4.02 (3H, s, CH <sub>3</sub> ON), 6.65 (1H, s, isoxazol), 10.00 (1H, d, CONH)
11d	OCH₃	79	81~83	$C_{17}H_{18}N_4O_9S$	2.07 (3H, s, $CH_3CO$ ), 3.98 (3H, s, $CH_3O$ ), 4.04 (3H, s, $CH_3ON$ ), 6.55 (1H, s, isoxazol), 10.00 (1H, d, $CONH$ )
12a	Br	76	88~90	$C_{16}H_{15}BrN_4O_8S$	2.05 (3H, s, CH <sub>3</sub> CO), 4.16 (3H, s, CH <sub>3</sub> ON), 7.51 (1H, s, isoxazol), 9.80 (1H, d, CONH)
12c	CH3	68	73~75	$C_{17}H_{18}N_4O_8S$	2.08 (3H, s, CH <sub>3</sub> CO), 2.33 (3H, s, CH <sub>3</sub> ), 4.13 (3H, s, CH <sub>3</sub> ON), 7.10 (1H, s, isoxazol), 9.70 (1H, d, CONH)

Table 4. 7-Aminodesacetoxycephalosporanic acid derivatives 10a, 10b and 7-aminocephalosporanic acid derivatives 11a~11d, 12a and 12c.

<sup>a</sup> Trituration solvent: *n*-Pentane.

<sup>b</sup> HPLC study showed all compounds had a purity >95%.

Compound	<i>S.a.</i> 209P	S.a. Z2ª	<i>E.f.</i>	S.p. A-32	<i>K.p.</i> 381 <sup>b</sup>	<i>E.c.</i> 128 <sup>b</sup>	<i>E.c.</i> 078	E.cl. 122 <sup>b</sup>	P.m. Vi	S.f. <sup>b</sup>
10a	64	32	>128	4	>128	>128	>128	>128	8	>128
10b	32	32	>128	1	>128	>128	>128	>128	>128	>128
11a	2	1	>128	0.25	32	16	8	16	1	8
11b	2	2	>128	0.125	>128	128	64	128	16	128
11c	2	2	>128	0.5	16	16	8	16	0.5	8
11d	4	4	>128	0.5	32	16	16	32	2	16
12a	16	16	>128	4	>128	>128	>128	>128	64	>128
12c	16	32	>128	2	>128	>128	>128	>128	16	>128
Cefuroxime	2	2	>128	0.0312	4	4	4	4	0.25	4

Table 5. In vitro antibacterial activity of compounds 10a, 10b, 11a~11d, 12a and 12c (MIC, µg/ml).

Test organisms and abbreviations: S.a. 209P, Staphylococcus aureus 209P; S.a. Z2, Staphylococcus aureus Z2; E.f., Enterococcus faecalis ATCC 10541; S.p. A-32, Streptococcus pyogenes A-32; K.p. 381, Klebsiella pneumoniae 381; E.c. 128, Escherichia coli 128; E.c. 078, Escherichia coli 078; E. cl. 122, Enterobacter cloacae 122; P.m. Vi, Proteus mirabilis Vi; S.f., Shigella flexneri 31172.

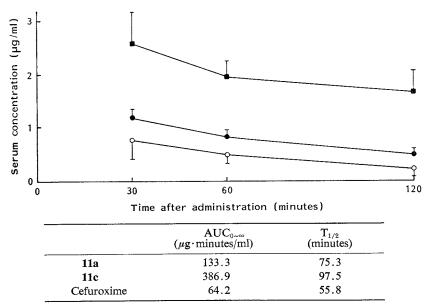
<sup>a</sup> Penicillinase producer.

<sup>b</sup> Cephalosporinase producer.

active against  $\beta$ -lactamase producer microorganisms. However, the nature of the substituent in position 3 at the isoxazole ring influenced significantly the activity. In fact, the presence of Br (11a) and CH<sub>3</sub> (11c) led to derivatives with significant antibacterial activity, whereas the presence of OCH<sub>3</sub> (11d) displayed a reduction in the activity against both Gram-positive and Gram-negative microorganisms. The introduction of Ph (11b) resulted in the decrease of the activity, especially against Gram-negative bacteria.

Fig. 1. Mean serum concentrations ( $\pm$ SE), AUC<sub>0~∞</sub> and T<sub>1/2</sub> of compounds 11a, 11c and cefuroxime after oral administration (100 mg/kg) in rats.





Our studies confirmed the importance of (Z)-isomerism of the oximino group of derivatives **11a~11d**; in fact, when (E)-isomers **12a**, **12c** were tested *in vitro*, a remarkable reduction of the activity was found. The replacement of the 7-aminocephalosporanic nucleus of compounds **11a~11d** with the 7-aminodesacetoxycephalosporanic nucleus of derivatives **10a**, **10b** resulted again in a decrease of the *in vitro* antibacterial properties. While the *in vitro* antibacterial profile of our compounds generally denoted less potency than cefuroxime, the oral bioavailability of derivatives **11a**, **11c** was found higher than that shown by the reference drug. The mean serum concentration time curves of compounds **11a**, **11c** and cefuroxime, after oral administration in rats, are reported in Fig. 1. In the same figure the values of area under the curves (AUC<sub>0~∞</sub>) and elimination half-life (T<sub>1/2</sub>) of these three compounds are also described. Particularly, derivative **11c** produced higher levels in serum, higher AUC<sub>0~∞</sub> and longer T<sub>1/2</sub> than those found in compound **11a** and cefuroxime.

#### Experimental

Melting points were determined on a Buchi SMP-20 and are uncorrected; boiling points are also uncorrected. <sup>1</sup>H NMR spectra were recorded on a Varian EM-360L spectrometer using TMS as internal standard. Solutions were dried over sodium sulfate and solvents were removed by evaporation under reduced pressure using a rotatory evaporator.

#### **Biological Methods**

Minimun inhibitory concentrations (MICs) were determined by the broth dilution method in microtiter trays, as already described<sup>8)</sup>, with Mueller-Hinton broth as the test medium, after incubation for 20 hours at 37°C with an inoculum size of about 10<sup>6</sup> cfu/ml. The serum concentrations of compounds **11a**, **11c** and cefuroxime were measured after oral administration, in rats (n=3), of 100 mg/kg of drugs. The serum samples were assayed by a disc-plate diffusion method, using *Bacillus subtilis* NCIB 8993 as test organism. The test medium was Lab-Lemco (Oxoid) supplemented with NaCl 0.5% and sodium citrate 0.25%. Standards were prepared with pooled normal rat antibiotic-

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free serum. The plates were incubated overnight at 37°C. Standards, controls and samples were run at least four times. The sensitivity of the assay was 0.25  $\mu$ g/ml for compounds 11a and 11c, while in the case of cefuroxime the lowest assayable concentration was 0.1  $\mu$ g/ml.

#### 2-(3-Bromo-5-isoxazolyl)-2-hydroxyacetic Acid Ethyl Ester (4a)

A solution of bromine (33.6 g, 0.21 mol) in AcOH (70 ml) was added dropwise at 60°C to a stirred solution of 3-bromo-5-acetylisoxazole (1a, 19.0 g, 0.1 mol) in AcOH (190 ml). After 10 minutes at 60°C, the solvent was evaporated; the residue was taken up with water (100 ml) and treated at 0°C with a solution of NaOH (20.0 g, 0.5 mol) in water (100 ml). The solution was stirred at room temp for 2 hours, neutralized with AcOH and extracted with ethyl ether ( $3 \times 100$  ml). The aqueous layer was then acidified with concd HCl to pH 2 and extracted with ethyl ether ( $3 \times 100$  ml). The organic layer was dried and evaporated to give an oily residue (19.0 g) which was stirred at room temp for 20 hours in EtOH (200 ml) with a catalytic amount of concd H<sub>2</sub>SO<sub>4</sub>. The solvent was removed and the residue was poured into water (150 ml) and extracted with ethyl ether (300 ml). The organic layer was washed with 10% aq NaHCO<sub>3</sub>, dried and evaporated. The solid residue was crystallized from isopropyl ether to give pure 4a as a white solid (16.2 g, 65%, mp  $50 \sim 51^{\circ}$ C).

Compounds  $4b \sim 4d$  were similarly prepared.

2-(3-Bromo-5-isoxazolyl)-2-oxoacetic Acid Ethyl Ester (5a)

To a stirred solution of 4a (21.5 g, 0.086 mol) in AcOH (300 ml) was added portionwise  $CrO_3$  (10.3 g, 0.103 mol) at room temp. The mixture was kept at 100°C for 2 hours; the solvent was removed and the residue was treated with water (300 ml) and neutralized with NaHCO<sub>3</sub>. The aqueous mixture was extracted with ethyl ether (3×200 ml); the organic layer was washed with water (2×50 ml), dried and evaporated. The residue was purified by distillation to afford pure 5a as a colorless oil (17.1 g, 80%, bp 115°C/0.3 Torr).

Compounds  $5b \sim 5d$  were similarly prepared.

2-(3-Bromo-5-isoxazolyl)-(Z)-2-methoxyiminoacetic Acid Ethyl Ester (6a) and <math>2-(3-Bromo-5-isoxazolyl)-(E)-2-methoxyiminoacetic Acid Ethyl Ester (7a)

A mixture of **5a** (14.9 g, 0.06 mol) and methoxyamine hydrochloride (5.3 g, 0.063 mol) in EtOH (200 ml) and water (200 ml) was refluxed for 2 hours. After evaporation of the solvent, the residue was treated with water (200 ml) and extracted with ethyl ether ( $3 \times 150$  ml). The organic layer was washed twice with water, dried and evaporated. The two isomers were separated by column chromatography (silica gel 1,000 g, eluent hexane - isopropyl ether, 9:1) to give pure **6a** (5.8 g, 35%, bp 135°C/0.2 Torr) and pure **7a** (9.3 g, 56%, bp 140°C/0.15 Torr).

Compounds  $6b \sim 6d$  and  $7b \sim 7d$  were similarly prepared.

#### 2-(3-Bromo-5-isoxazolyl)-(Z)-2-methoxyiminoacetic Acid (8a)

To a stirred solution of **6a** (5.54 g, 0.02 mol) in MeOH (30 ml) was added dropwise at room temp a solution of NaOH (0.88 g, 0.022 mol) in water (30 ml). The mixture was stirred at room temp for 2 hours; the solvent was removed and the residue was treated with water (80 ml) and extracted with ethyl ether ( $2 \times 30$  ml). The aqueous layer was acidified with concd HCl to pH 2 and extracted with ethyl ether ( $3 \times 50$  ml) which was dried and evaporated. The solid residue was crystallized from benzene (30 ml) to give pure **8a** as a white solid (4.68 g, 94%, mp 136~137°C).

Compounds  $8b \sim 8d$  and  $9a \sim 9d$  were similarly prepared.

3-Acetoxymethyl-7-[2-(3-bromo-5-isoxazolyl)-(Z)-2-(methoxyimino)acetamido]-8-oxo-5-thia-1azabicyclo[4.2.0]oct-2-ene-2-carboxylic Acid (11a)

80% NaH dispersion in oil (0.15 g, 0.005 mol) was added portionwise at 5°C to a stirred solution of 8a (1.24 g, 0.005 mol) in benzene (25 ml) and DMF (2 ml). The mixture was stirred at room temp for 1 hour and oxalyl chloride (1.5 ml, 0.010 mol) was then dropped at 5°C.

The mixture was stirred at room temp for 1 hour; the solvent was removed and the residue, dissolved in THF (20 ml), was dropped to a solution prepared refluxing for 1 hour a mixture of 7-aminocephalosporanic acid (1.63 g, 0.006 mol) and hexamethyldisilazane (1.25 ml, 0.006 mol) in THF (25 ml) and adding after cooling, triethylamine (0.84 ml, 0.006 mol). The resulting mixture was stirred at room temp for 3 hours, evaporated and poured into a cold 5% aq NaHCO<sub>3</sub> (50 ml). The aqueous layer was washed with ethyl acetate (100 ml), acidified with 10% HCl to pH 3 and extracted with ethyl acetate ( $2 \times 50$  ml). The organic layer was washed with water, dried and evaporated. The residue was triturated with *n*-pentane (50 ml) to give a practically pure solid **11a** (1.71 g, 68%, dec point 70~72°C).

Compounds 10a, 10b, 11b~11d, 12a and 12c were similarly prepared.

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